NATIONAL BRAIN RESEARCH CENTRE(NBRC) NH-8, Manesar-122050, HARYANA

Sample Questions For Ph.D. Entrance Examination-2011

<u> </u>	app	ear in the entrance stions in the entrance	test. Th	•	•	•		
<u>PI</u>	HYSIC	<u>:S</u>						
1.	Name the process or processes which alternate (s) the passage of gamma rays through matter:							
	(a) (c)	Photoelectric effect Pair Production	(b) (d)	Compton Recoils All of these				
2.	How many quantum numbers do characterise an electron within an atom:							
	(a) (c)	Six Four	(b) (d)	Five Three				
3.	Write down the total number of particles inside the nucleus in terms of mass number(A and atomic number Z:							
	(a) (c)	A–2Z 3A–2Z	(b) (d)	2A–Z None of these				
4.	What is the name of the detecting device for discovery of positron:							
	(a) (c)	Electron Microscope Bubble Chamber	(b) (d)	Geiger-Muller Cour Cloud Chamber	nter			
5.	Give the name of the process by which a positron and an electron coalesce to form photons:							
	(a) (c)	Pair Annihilation Bremhausstrahlung	(b) (d)	Compton Scattering Rutherford Scatteri				
<u>CF</u>	<u>IEMIS</u>	<u>TRY</u>						
6.	When solid sodium chloride crystallises from a saturated solution, the entropy change of the system:							
	(a) (c)	Decreases Zero	(b) (d)	Increases Remains same with	n the surround	ding		
7.	Pure water has a $P_H = 7$ at 25° C. The water is heated to 90° C. Choose the right answer:							
	 (a) K_w value will decrease and water will be alkaline (b) K_w Value will decrease but water will remain neutral (c) K_w Value will increase and water will be acidic (d) K_w value will increase but water will remain neutral 							

	8.	The vapour pressure of pure components 'X' and 'Y' are 100 torr and 50 torr respectively at a particular temperature. Assuming solution of X & Y obeying Raoult's law, the mole fraction of 'X' in the vapour phase in equilibrium with a solution containing equimoles of 'X' and 'Y' is:									
		(a) (c)	0.33 0.80	(b) (d)	0.66 0.50						
	9.	The half life time for a reaction at initial conc. of 0.1 mole/litre and 0.4 mole/litre are 200 sec and 50 sec respectively. The order of the reaction is:									
		(a) (c)	0 2	(b) (d)	1						
	10.		equilibrium condition he value of K _p is:	on for the rea	action 2C(s)+O ₂ (g)	⇒ 2CO(g) the total pressure is 12					
		(a) (c)	16 2	(b) (d)	0.5 32						
	BIC	<u>DLOG</u>	<u>Y</u>								
	11. The salivary gland of leech secrete the anticoagulating factor:										
		(a) (c)	Hypotoxin Hirudin	(b)	Antitoxin None of these						
	12.	The be	eak is toothed in:								
		(a) (c)	Ostrich Penguin	(b) (d)	Archaeopteryx Kiwi						
	13.		ykol is:	/	7)						
		(a) (c)	A hormone A pheromon	(b) (d)	An enzyme None of these						
	14.	The major constituent of blood plasma (more than half) is:									
		(a) (c)	Protein Fatty substance	(b) (d)	Inorganic salts Sugar						
	15.		Il Ring indicates:								
		(a) (c)	Age of the tree Scale bark	(b) (d)	Branching point Adventitious root						
	M	ATHE	EMATICS								
16.	If 7	If T is a real, skew-symmetric n×n matrix, then for n odd:									
	(a) (c)				(b) det T < o(d) None of thes	se					
17.	Th	The smallest subgroup which contains a fixed open subset of a topological group is:									
	(a) (c)	-	open th open and closed		(b) Is only close (d) All of these a						
18.	A subset S of real numbers has at least one limiting point if:										
	(a) (c)		oounded oounded and finite		(b) S is infinite s (d) S is bounded						

- 19. Let F be a field of real numbers and $\begin{pmatrix} 0 & 1 \\ -1 & 0 \end{pmatrix} \in F_2$, if D be the set of all matrices commuting with $\begin{pmatrix} 0 & 1 \\ -1 & 0 \end{pmatrix} F_2$, then:
 - (a) D is isomorphic to the field of complex numbers (b) D is isomorphic to the field of real numbers
- 20. If T is a linear transformation on an inner product space V(F), then which of the following is incorrect:
 - (a) If T is self –adjoint, then Tr T is real

(c) D is not isomorphic to the field of complex numbers (d)

(b) If $T \ge 0$ then $Tr T \ge 0$

(c) If T >o then Tr T <o

(d) Not all the above are ture

None of these

