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2009-2010
B.Sc. (Hons.) (PART – III) EXAMINATION
(PHYSICS)
ATOMIC, MOLECULAR, LASER AND SOLID STATE PHYSICS
(PH-309)

Maximum Marks: 40

Duration: Three Hours

Note: Answer all questions.

1. (a) Using vector model, obtain the terms and levels of three-electron configuration spd. Explain the order of their occurrence in the light of Hund's rule. (4)
- (b) Explain the fine structure of energy level of Sodium atom. Describe the electric dipole selection rules for the observed transitions. (3)
- OR
- 1' (a) Explain the effect of spin-orbit interaction energy on the term system. Derive an expression for spin-orbit interaction energy. (5)
- (b) Calculate the fine structure splitting of 2D term. (2)
- 3) 2 (a) Describe the rotation-vibration spectrum of a diatomic molecule. What are the selection rules to observe O,P,Q,R and S branches? (4)
- (b) Describe the formation of bonding and antibonding of 2p orbitals in accordance with molecular orbital theory. Draw the molecular orbital energy level diagram of a diatomic molecule. (3)
- OR
- 14) (b') Describe the pure rotational Raman spectrum of a diatomic molecule. Why Stokes lines are more intense than Antistokes lines? (3)
- 3 (a) Show that the probability of induced emission is same as the probability of induced absorption. Explain the role of metastable states in lasers. (3)
- (b) With suitable energy level diagram, explain the working of ruby laser. Explain the process of optical pumping used to achieve population inversion. (3)
- 4 (a) What are symmetry operations? Describe any two symmetric operations. (4)
- (b) Discuss the origin of Van-der waals bonding. (3)
- OR
- 4' (a) What are the various types of dislocations in a crystal? Discuss the role of dislocations in crystal growth. (5)
- (b) Show that the spacing d_{111} is approximately 1.2 times the spacing d_{200} in Pb that has fcc structure. The diameter of Pb atom is 1.8\AA . (2)

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- 5 (a) Discuss the role of electron and neutron diffraction in the structural analysis of crystals, mentioning the points where their application outstands from X-ray diffraction. (3)
- (b) Discuss the main features of acoustical and optical branches obtained in the vibrational spectrum of a linear diatomic chain. (3)
- OR
- (b') Calculate the angle of incidence at which electrons of energy 100eV must be incident on the lattice planes of a metal crystal in order to give a strong Bragg reflection in the first order, given that the lattice spacing is 2.5 Å. (3)
- 6 (a) What is Hall effect? Mention the important applications of this effect. (3)
- (b) Calculate the band gap for a linear mon atomic crystal using nearly free electron model. (4)

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