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Total No. of Questions: 0	91

[Total No. of Pages: 02

# Paper ID [CH101]

(Please fill this Paper ID in OMR Sheet)

B.Tech. (Sem. - 1st/2nd)

CHEMISTRY (CHM - 101)

Time: 03 Hours

Maximum Marks: 60

### Instruction to Candidates:

- 1) Section A is Compulsory.
- 2) Attempt any Five questions from Section B & C.
- 3) Select at least Two questions from Section B & C.

#### Section - A

01)

 $(10 \times 2 = 20)$ 

- a) Softening of water is essential, Explain.
- b) Differentiate between temporary and permanent hardness.
- c) What is pitting corrosion? Explain.
- d) Explain the corrosion taking place in acidic medium.
- e) Draw the phase-Diagram of water-system.
- f) Explain the laws of photochemistry.
- g) Give any one method for the prevention of corrosion.
- h) What is meant by standard electrode potential? Explain.
- i) Explain the NMR spectrum of acetone, methyl alcohol and benzene.
- j) Define chromophore and explain with examples.

## Section - B

(Marks: 8 Each)

- Q2) What do you understand by cation and anion exchangers? Explain the removal of hardness with ion exchange method.
- Q3) Give the mechanism of metal corrosion in slightly basic medium. Give some specific examples.

R-2023 [2058]

- Q4) Explain the Lowry-Bronsted and Lewis concept of acid and base. Whethe BF<sub>3</sub> and pyridine is acid or base, explain.
- Q5) Write short notes on the following:
  - (a) Cathodic protection.
  - (b) Atmospheric Corrosion.

#### Section - C

(Marks: 8 Each)

- Q6) Explain the Law of Photochemical equivalence. How quantum yield can be explained taking Law of Photochemical equivalence into consideration. Explain with some specific examples.
- Q7) What do you mean by chemical-shift? Explain with examples the shielding and deshielding effects of halogens in case of an alkane.
- Q8) Define Phase, component and degree of freedom. Discuss Phase-Diagram for Liquid-Solid System.
- Q9) Write short notes on the following:
  - (a) Transitions involved in UV-Spectrophotometry.
  - (b) Photoelectron Spectroscopy in case of atoms and molecules.

