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Printed Pages: 2 PH – 121

(Following Paper ID and Roll No. to be filled in your Answer Book)

PAPER ID: 5046

Roll No.

## B. PHARM.

## (SEM. II) EXAMINATION, 2006-07 PHYSICAL CHEMISTRY

Time: 3 Hours | [Total Marks: 80]

**Note**: (i) Answer all the questions.

- (ii) All questions carry equal marks.
- 1 Attempt any **four** of the following: 4×4
  - (a) What are the assumptions of kinetic molecular theory of gases ?
  - (b) Derive van der Waals equation of state for Real Gases
  - (c) Define surface tension. What methods are employed to determine surface tension?
  - (d) Describe briefly ideal and real solutions.
  - (e) Define Molal Depression Constant. Write a method to determine the molecular weight of non volatile solute.
- 2 Answer any **four** of the following: 4×4
  - (a) State and explain first law of thermodynamics
  - (b) Distinguish between:
    - (i) Isothermal and adiabatic process
    - (ii) Reversible and Irreversible process.
  - (c) Describe briefly Joule Thomson Effect.

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- (d) Define: Adsorption, Absorption, Adsorbate, Adsorbent.
- (e) Describe Langmuir adsorption isotherm.
- 3 Answer any **four** of the following: 4×4
  - (a) Define pH. What are the methods for determining pH of a solution?
  - (b) Derive an equation for First Order Reaction.
  - (c) State and explain Kohlrausch Law.
  - (d) Write a note on: Complex Reactions
  - (e) Describe Homogeneous and heterogeneous catalysis.
- 4 Write notes on any **four** of the following:  $\sqrt{4}$  4×4
  - (a) Heat of Formation
  - (b) Enthalpy of solution and Enthalpy of Hydration.
  - (c) Heat of Combustion
  - (d) Heat of Neutralization
  - (e) Exothermic and Endothermic Reactions.
- 5 Answe any four of the following: 4×4
  - (a) State Phase Rule. What are its applications over water system ?
  - (b) Describe the validity of distribution law in case of association or dissociation of solute in one of the phases.
  - (c) What are the different types of crystal systems?
  - (d) Write brief notes on:
    - (i) Partition coefficient
    - (ii) Congruent and Incongruent melting point
    - (iii) What are the features of a crystal?

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