Dr.R.M.L.Avadh University

Mark- 50

B.Sc. IIIrd year

Year 2008

Paper I: General Chemistry, Molecular Spectroscopy and Analytical chemistry

Note-Attempt Question No. 1 which is compulsory and four more from the rest .Marks allotted to each question are given against it.

- 1- A) What are common toxic elements? Discuss their effects in the environment, 1-mark
 - B) Discuss the number of different symmetry elements in benzene molecule. 1-mark
 - C) Explain how monuments are damaged by pollutants.
 - D) Explain how some nuclear disintegrations become self-sustaining. 1-mark
 - E) How do kinetic methods determine the reaction mechanism? 1-mark
 - F) State Born-Oppenhiemer approximation. 1-mark
 - G) How is I.R spectroscopy employed for watching the progress of a chemical change? 1-mark
 - H) What is isotope effect in rotational spectra? 1-mark
 - I) Is melting point true criteria of purity of a substance? 1-mark
 - J) Explain in brief the shielding and deshielding of protons. 1-mark
 - 2. What is electromagnetic radiation? Describe the various regions of electromagnetic radiation concerned with the corresponding spectroscopic techniques. 10-mark
 - 3. Discuss the rotational spectrum of a diatomic rigid rotor. How is the bond-length of a molecule determined by it? 10-mark
 - 4. A) Write down the types of electronic transitions and selection rules for d-d transitions.5-mark
 - B) Discuss the electronic spectrum of $[Ti(H_2O)_6]^3$ complex. 5-mark

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- 5. A) What is the basic principle of thin layer chromatography? Describe its applications. 5-mark
 - B) State Lambert-Beer" law and give the applications of Colorimetric analysis. 5-mark
 - 6. What are Raman spectra? Discuss the origin of Raman lines. 10-mark
 - 7. What are radio isotopes? How are they separated? Give their applications, 10-mark
 - 8. What are the symmetry operations? Explain them with suitable examples. 10-mark
 - 9. What do you mean by chemical shift? Discuss the PMR spectra of ethyl bromide. 10-mark
 - 10. Write short notes on any four of the following-2.5+2.5+2.5+2.5=10 mark
 - A) Photochemical pollutants
 - B) Degree of freedom
 - C) Polarisibility of molecule
 - D) Franck -Condon Principle
 - E) Polarography and its applications.

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